UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

MARK SCHEME for the May/June 2012 question paper for the guidance of teachers

5129 COMBINED SCIENCE

5129/02

Paper 2 (Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2012 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

	Pa	ge 2	Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL – May/June 2012	5129	02
1	(a)	(i) t	transport/carry oxygen ;		[1]
		r	presence of haemoglobin ; no nucleus present ; large surface area (per volume)/biconcave disc shape	any 2	[max 2]
		r r	combines with oxygen for transport/releases oxygen in more haemoglobin contained within the cell/more oxygmore oxygen can pass into the cell (in lung capillaries), cell (in tissue capillaries);	en carried ;	leave the
		ć	adaptation and explanation must be linked correctly		[max 2]
	(b)	plasr	<u>ma</u> ;		[1]
2	(a)		ma or a = F/m or 0.32/0.2 ;		
		1.6 ; m/s²			[3]
	(b)	2;			[1]
3	(a)	71 ; 7.1 ; 2.92	117; 11.7; (divided by 10) 5g (divided by 4) (ecf throughout)		[4]
	(b)	ionic	/electrovalent ;		[1]
	(c)	<u>kill</u> ba	acteria/micro-organisms/germs ;		[1]
4	(a)	Nm ;			[1]
	(b)		e applied further from fulcrum (pivot)/perpendicular distaller force gives same moment/larger moment for same		[2]
5	(a)	1 (r	elative charge) ; elative mass) ; elative charge) ;		[3]
	(b)	numl	ber of neutrons/number of nucleons/mass number ;		[1]
	(c)	same	e number of electrons in outer shell ;;		[2]

Mark Scheme: Teachers' version

Syllabus

Paper

	Page 3		}			hers' version	Syllabus	Paper
				GCE O LE	VEL – Ma	ay/June 2012	5129	02
6	(a)	(i)	B, C	or D ;				[1]
		(ii)	<u>E</u> ;					[1]
	(b)	produces (hydrochloric) acid; kills bacteria (on food)/prevents food poisoning;						
		or acc	ent st	tores food ;				
		no need to eat constantly/can concentrate on other activities;						[max 2]
	(0)	hilo	woul	ld not be added (to	the food h	oing digostod):		[1]
	(C)	the	fat (ir	Id not be added (to n the food) not emu	Isified;			[1]
		sto	mach	tion would be incon acid would not be	neutralise	d ;	any 2	
		acti	on of	pancreatic enzyme	es impaire	d;		[max 2]
7	(a)	aer	obic ((respiration) uses o	xygen, an	aerobic does not ;	٦	
		aer	obic (respiration) release ic (respiration) prod	es more er	nergy than anaero	bic; any 2	
				produces carbon die			J	[max 2]
	(h)	hre	athin	g becomes more ra	nid/faster			
	(2)			g becomes deeper/	•		;	[2]
	(0)	/:\	1500	0/m):				[4]
	(6)	(i)		<u>0</u> (m) ;				[1]
		(ii)		onger the distance ept converse or cor			ic respiration ;;	[2]
8	(a)			mplitude ; vavelength ;				[2]
				0				
	(b)	v = 3.0		$\lambda = f/v \text{ or } 0.5 \times 6;$				[2]
		5.0	,					[2]
9	(a)	сор	per		→	reacts vigorously	with steam ;	
		ma	gnesi	um	*	reacts vigorously	/ with water ;	
		iror	1		\times	no reaction ;		
		pot	assiu	m	\	reacts slowly wit	h cold water and stea	m ; [4]
	(b)	<u>ligh</u>	ted/b	ourning splint explo	des with a	pop;		[2]

	Page 4	Mark Scheme: Teachers' version	Syllabus	Paper			
10	root hair		5129	02			
	osmosis ; xylem ; transpiration ;						
11		is positively charged ; e charges repel ;		[2]			
	(b) curr	ent;		[1]			
12		I = P/V or P = VI or 60/240; = 0.25;		[2]			
	(ii)	E = Pt or P = E/t or E = VIt or 60 × 600 ; = 36 000 ; (600 max 1 mark)		[2]			
	(b) (i)	microwave/radio;		[1]			
	(ii)	X-rays/gamma rays ;		[1]			
13	(a) met	hane ;		[1]			
	(b) com	npound of carbon and hydrogen only ;;		[2]			
	(c) 13	8 10 (all three);		[1]			
	(d) (i)	sulfur dioxide;		[1]			
	(ii)	acid rain ; corrodes buildings/kills plant or aquatic life ;		[2]			

	Page 5			Mark Scheme: Teachers' Version	Syllabus	Paper
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14	(a)	(i)	<u>Sun</u>	;		[1]
		(ii)	light	;		[1]
	(b)	(i)		erbivores) ; arnivores) ;		[2]
		(ii)	6 (sp	pecies);		[1]
		(iii)	exar	rgy lost at each trophic level; mple of energy loss (respiration, heat, digestion etc.) onger the food chain, the less energy (there is to pa); ss on);	any 2 [max 2]
	(c)		•	er population would decline/fall/less spiders ; eat moths/less food for spiders ;		
		mor		asshoppers/grasshopper population increases; wering plants/food for grasshoppers;		[max 2]
15	(a)	v = = 8		r 400/50 ;		[2]
	(b)	dire	ction	keeps changing/velocity is directional;		[1]
16	(a)	cop	•			[2]
	(b)	(i)	cutle	ery/chemical plant/surgical equipment/named exan	nples ;	[1]
		(ii)		ng metals /adding other elements to a metal to chaples of changing property;	nange/improve	properties ; [2]
17	(a)	ene	rgy c	an be neither lost nor created ;		[1]
	(b)	ther	mical mal/ etic ;	l ; heat ;		[3]

Mark Scheme: Teachers' version

Syllabus

Paper

Page 6		ge 6	Mark Scheme: Teachers' version	Syllabus	Paper
			GCE O LEVEL – May/June 2012	5129	02
18	(a)	calcium o	carbonate ;		[1]
	(b)	aluminiu	m oxide ;		[1]
	(c)	potassiu	m nitrate ;		[1]
	(d)	calcium o	carbonate ;		[1]
19	(a)	iron (core	e) ; and secondary (correct way round) ;		[2
	(b)	-	ging current/changing magnetic field (in iron core) ed e.m.f./voltage (in secondary)/current;	;	[2
20		e taken ; count rate	/activity/number of nuclei to halve ;		[2